DFT (B3LYP) COMPUTATIONAL STUDY ON THE MECHANISMS OF FORMATION OF SOME SEMICARBAZONES

Abdulfatai Siaka^{a*}, Adamu Uzairu^b, Sulaiman Idris^a, Hamza Abba^a

^aDepartment of Chemistry, Ahmadu Bello University, P.M.B 51, Zaria 810001, Kaduna, Nigeria ^bDepartment of Applied Chemistry, Federal University, P.M.B 5001, Dutsin-ma 823103, Katsina, Nigeria ^{*}e-mail: fatsaadaby@gmail.com

Abstract. Thermodynamic and kinetic mechanisms of forming six semicarbazones have been investigated computationally by DFT B3LYP method. All the reactions proceed *via* two transitions and include two consecutive steps: bimolecular and unimolecular. The computed transition steps have varying equilibrium constants values, enthalpy of activation and Gibbs energy of activation, depending on the semicarbazone involved. Also depending on the semicarbazones involved some of the consecutive steps are found to have varying enthalpy of reactions and spontaneity.

Keywords: semicarbazone, kinetics, bi- and unimolecular step, transition step, spontaneity.

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