

CRYSTAL STRUCTURE AND NMR SPECTROSCOPIC CHARACTERIZATION OF 1,5-BIS(2-HYDROXY-3-METHOXYBENZYLIDENE)CARBOONOHYDRAZIDE

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Abstract. The solid-state structure of a symmetrical carbohydrazone, namely 1,5-bis(2-hydroxy-3-methoxybenzylidene)carboono hydrazide (**1**) was determined by X-ray single crystal diffraction method. Compound **1** crystallizes in the monoclinic space group $P2_1/n$ with unit cell parameters $a = 10.1198(6)$, $b = 22.7847(11)$, $c = 15.1738(10)$ Å, $\beta = 100.458(6)^\circ$, $Z = 4$, $V = 3440.6(3)$ Å³, $R_1 = 0.0540$. Crystal structure of **1** is defined by two crystallographic independent molecules, which are bonded via N–H···O hydrogen bond. The organic molecules are as keto tautomers with respect to the carbamide fragment, and adopt the *anti* conformation. 1D and 2D NMR experiments have argued on the presence of the title compound in DMSO-*d*₆ solution mostly as keto tautomer in *syn* conformation, and enol-imino form when considering *o*-vanillin residue.

Keywords: carbohydrazone, *o*-vanillin, *syn-anti* isomer, X-ray diffraction, NMR spectroscopy.

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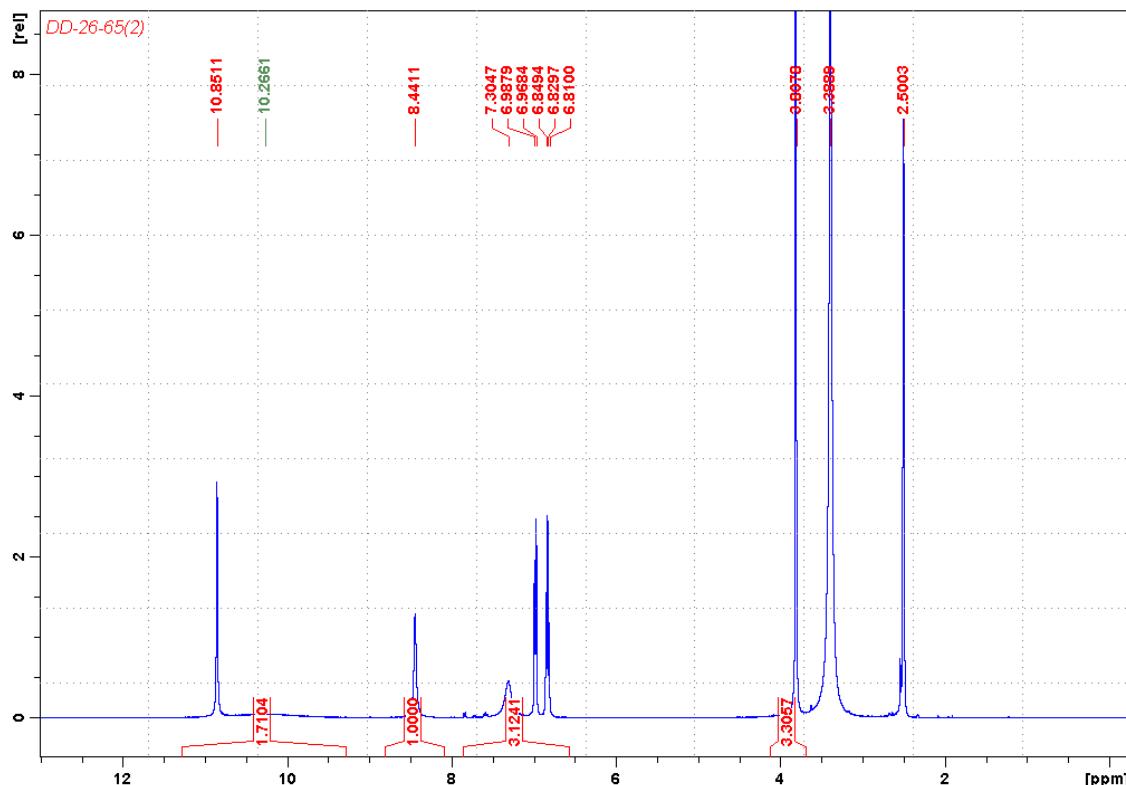


Figure S1. ¹H NMR spectrum of 1,5-bis(2-hydroxy-3-methoxybenzylidene)carboono hydrazide **1**.

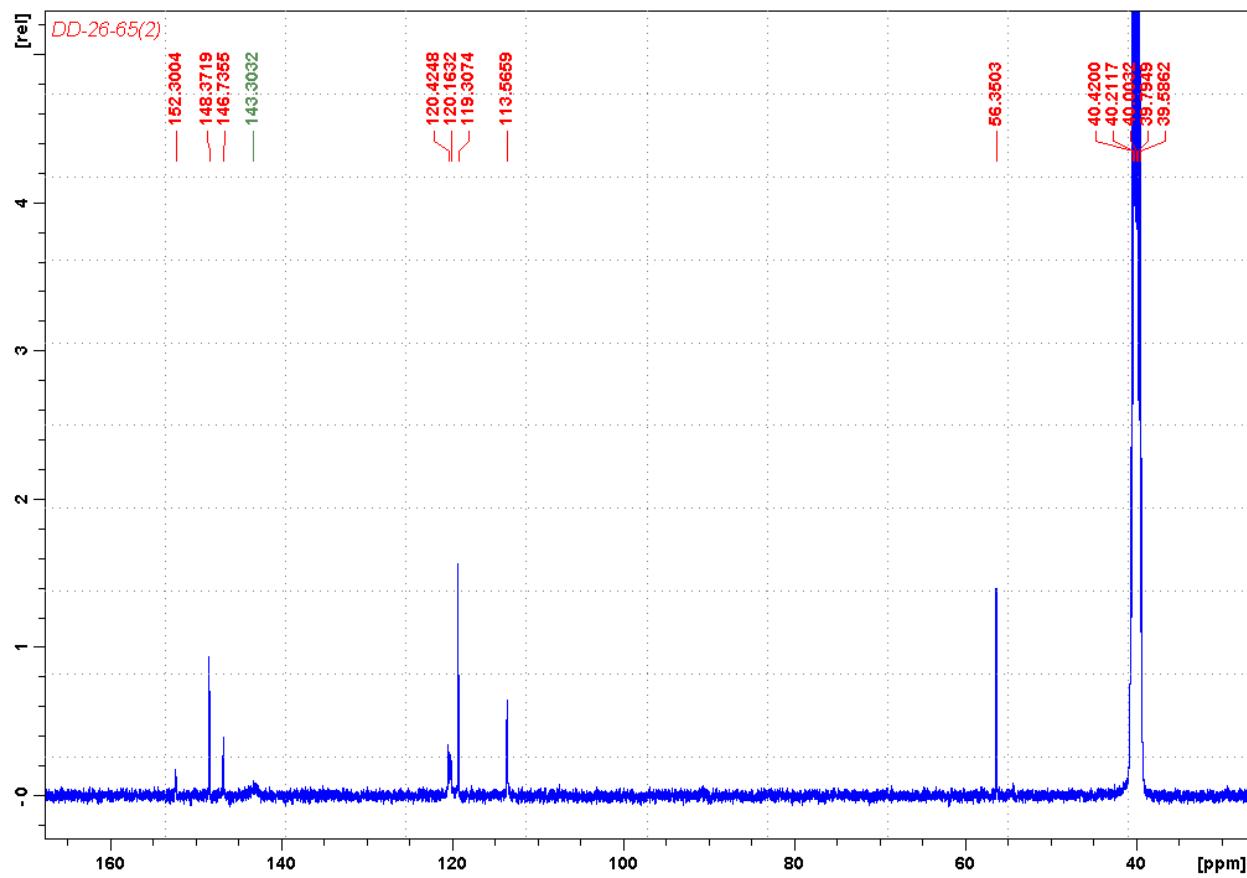
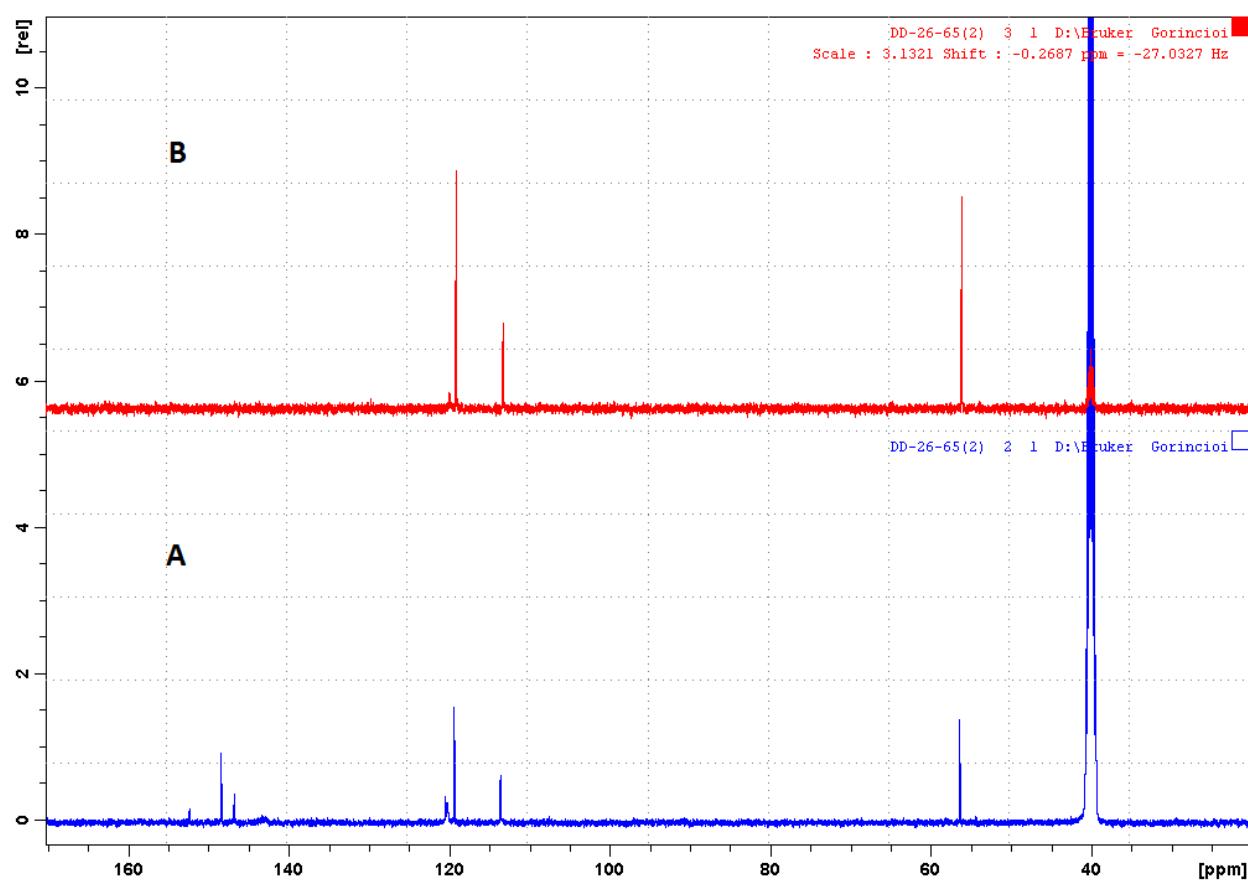
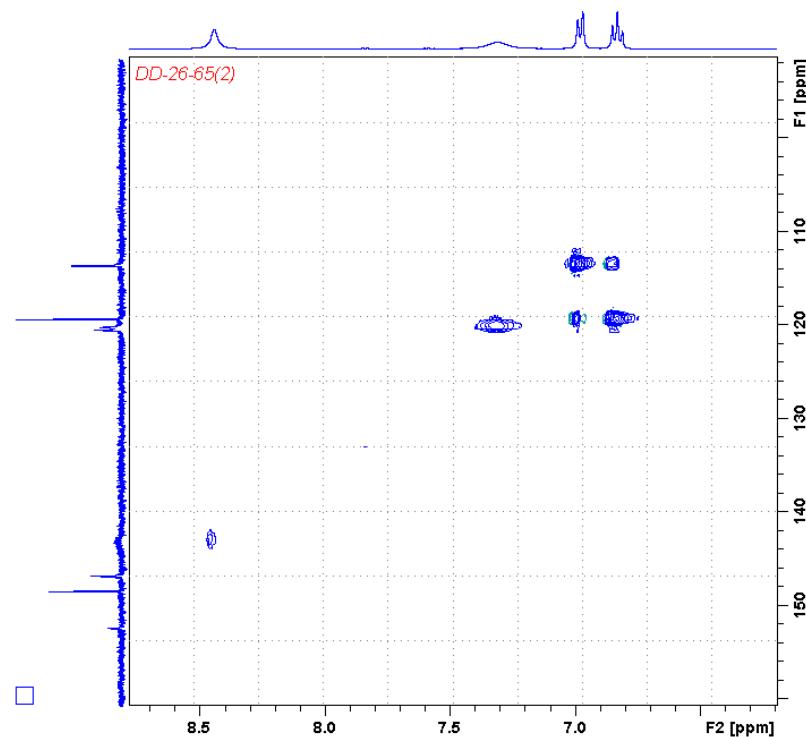


Figure S2. ^{13}C NMR spectrum of compound 1.

Figure S3. ^{13}C (A) and DEPT-135 (B) NMR spectra of compound 1.Figure S4. $^1\text{H}/^{13}\text{C}$ qHSQC NMR spectrum of compound 1.

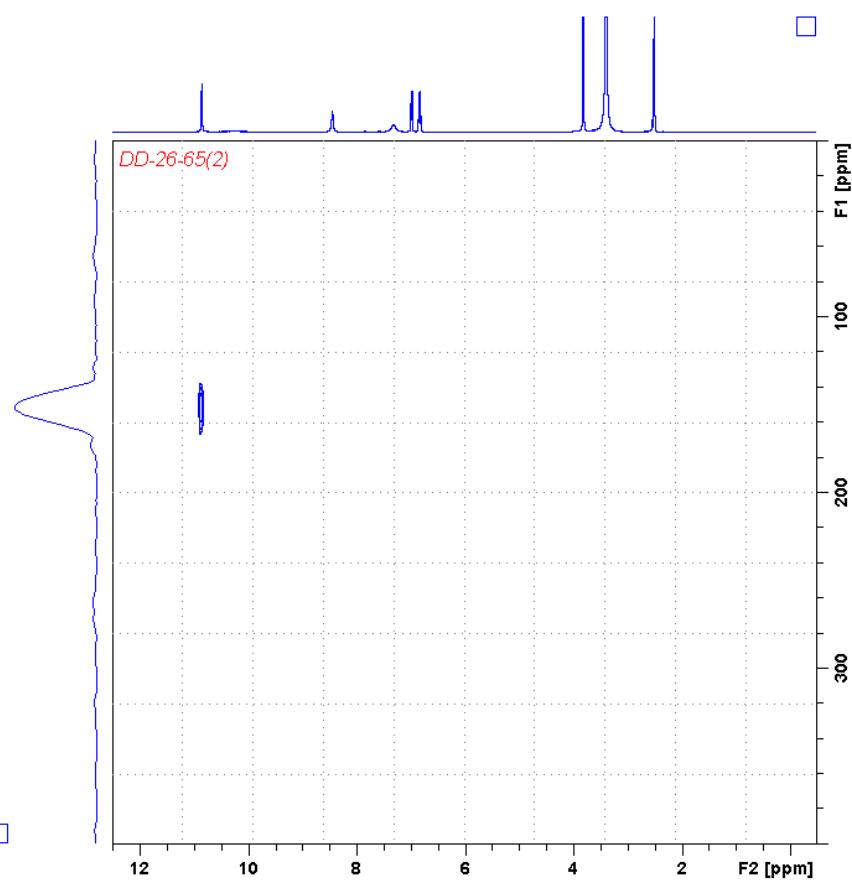


Figure S5. $^1\text{H}/^{15}\text{N}$ HMQC NMR spectrum of compound 1.

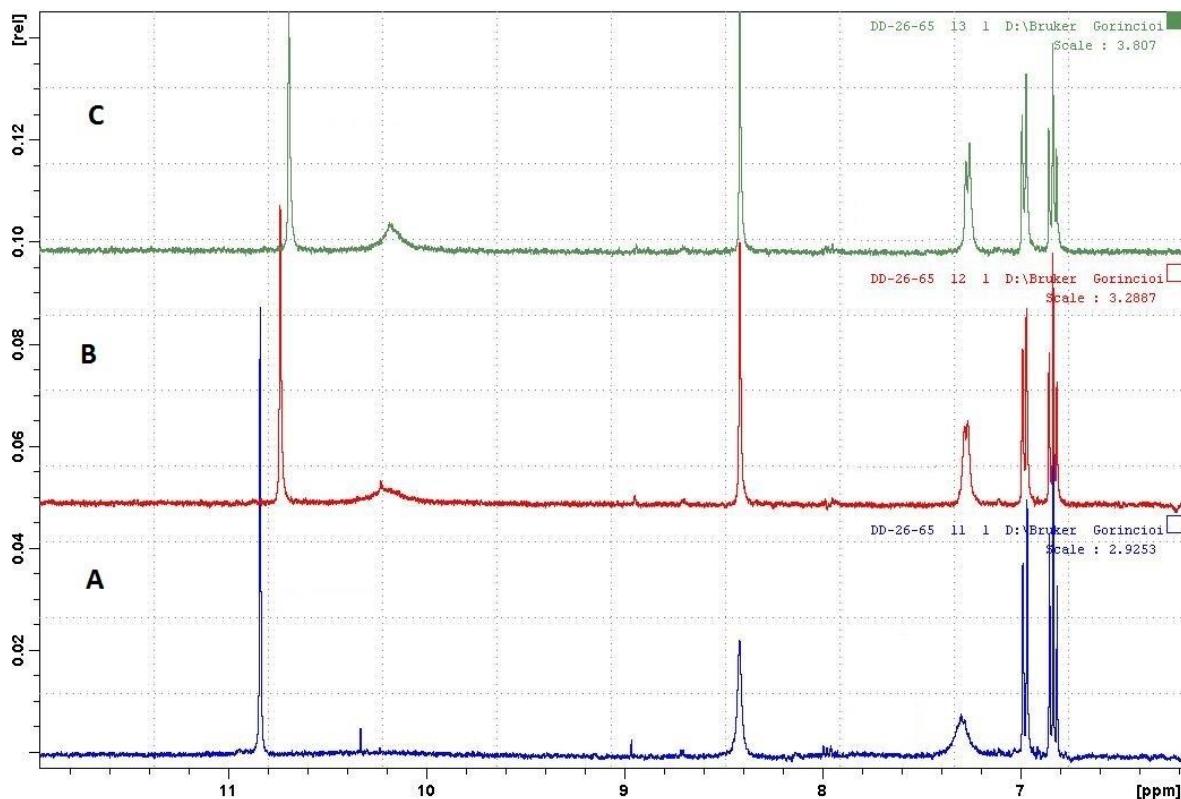


Figure S6. ^1H spectra of compound 1 recorded at three temperatures: A - 298K, B - 318K, C - 328K.

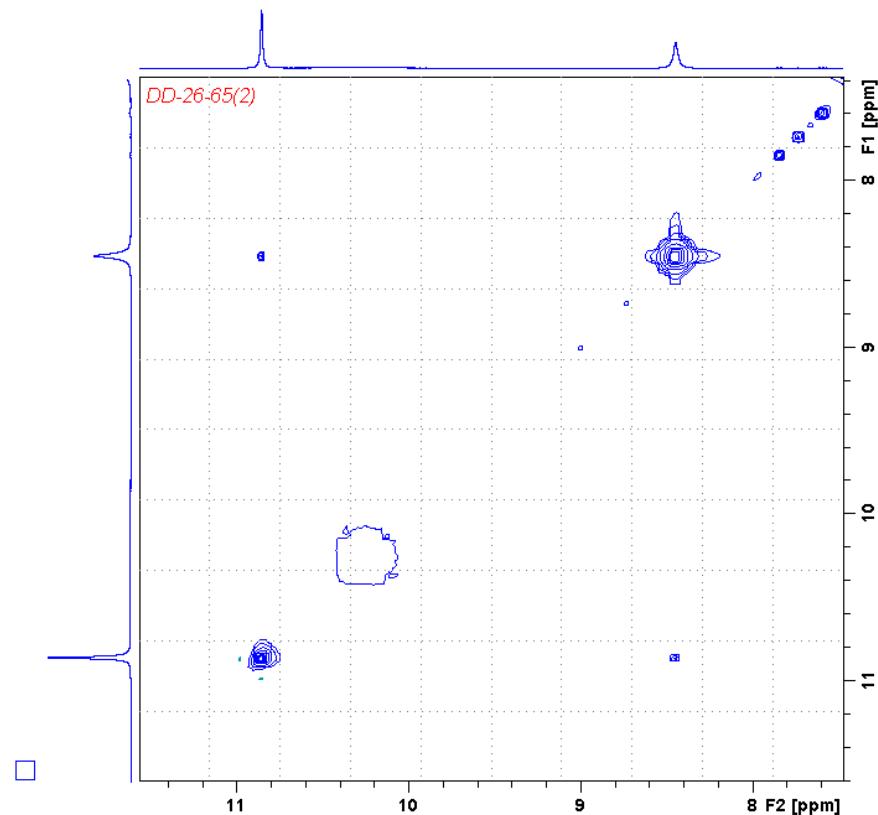


Figure S7. ¹H/¹H NOESY NMR spectrum of compound 1.

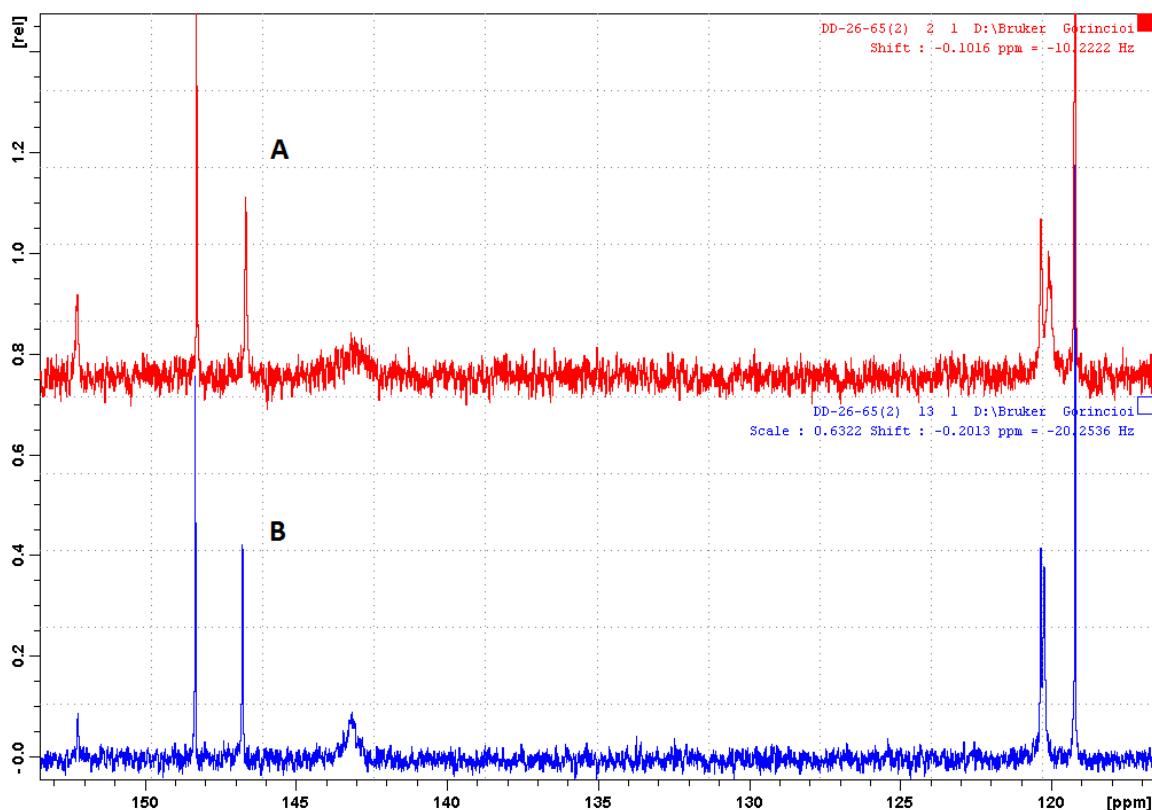


Figure S8. Fragment of ¹³C NMR spectra of compound 1 recorded at two temperatures: A- 298K, B- 348K.

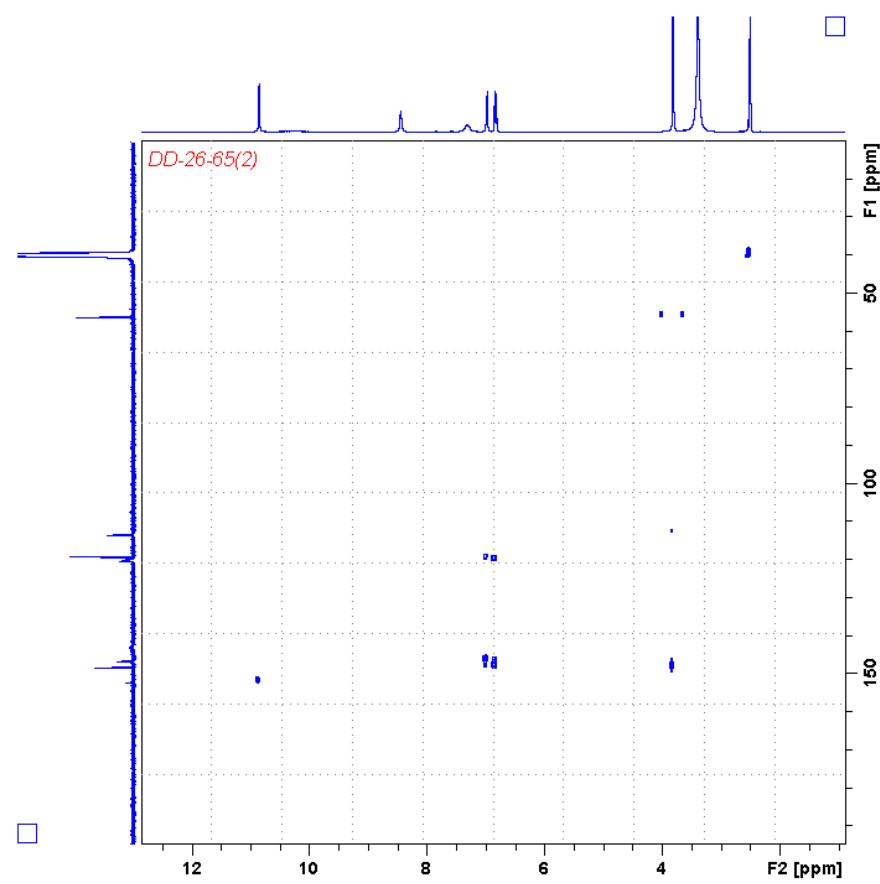


Figure S9. $^1\text{H}/^{13}\text{C}$ HMBC NMR spectrum of compound 1.

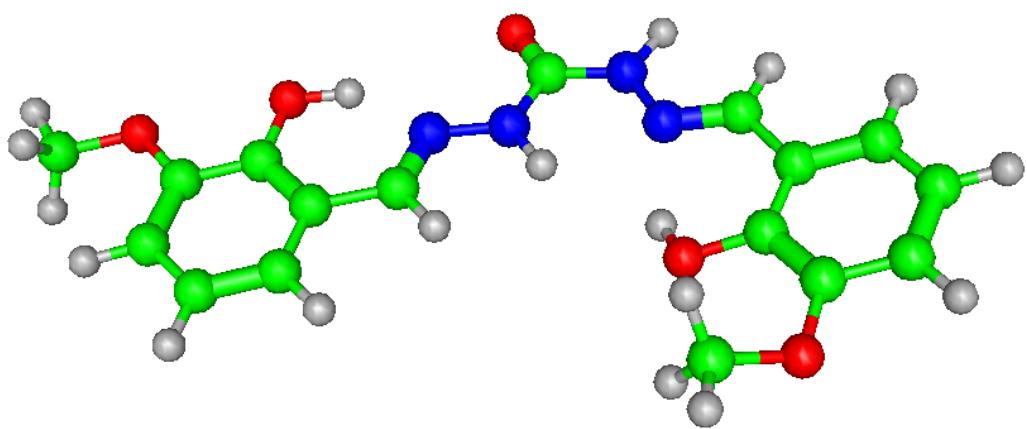


Figure S10. Minimum-energy stereo structure for an *anti*- conformer of compound 1 calculated by using PERCH NMR TOOLS (version 2014.1).