SYNTHESIS, CHARACTERIZATION, HIRSHFELD AND ADMET ESTIMATION STUDIES OF NOVEL 3-(2,4,6-TRIMETHYL-PHENYLAMINO)-BUT-2-ENOATE

Mohamed Loughzail^a, Koffi Senam Etsè^{ob}, Zaragoza Verez Guillermo^{oc}, Rachid Touzani^d, Anna Moliterni^e, Abdessamad Tounsi^{of}, Mohamed Anouar Harrad^{of,g*}

^aMolecular Chemistry Laboratory, Cadi Ayyad University, Marrakech 40000, Morocco ^bMedicinal Chemistry Laboratory, Center for Interdisciplinary Research on Medicines (CIRM), University of Liège, B36 Av. Hippocrate 15 B-4000 Liege, Belgium ^cX-Ray Diffraction Unit, RIAIDT, University of Santiago de Compostela, Campus VIDA, 15782 Santiago de Compostela, Spain ^dApplied Chemistry and Environment Laboratory, University of Mohamed 1, Oujda, Morocco. ^eInstitute of Crystallography Via Amendola, 122/O, Bari 70126, Italy ^fEnvironmental, Ecological, and Agro-Industrial Engineering Laboratory, Sultan Moulay Slimane University, Beni-mellal 23000, Morocco ^gRegional center for Education training and formation, CRMEF 40000 Marrakech-Safi, Morocco ^{*}e-mail: ma.harrad@yahoo.fr

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Table S1

Crystallographic data, details of data collection	a, and structure refinement parameters for compound 3.
Parameters	Value
Chemical formula	$C_{15}H_{21}NO_2$
Formula weight (g/mol)	247.33
Wavelength	0.71073 Å
Crystal system	Monoclinic
Space group	$P 2_{l}/c$
Temperature	100 K
Unit cell dimensions	a= 8.5003 (3) Å, b = 20.0682 (8) Å, c = 8.1720 (4) Å
	$\alpha = 90^{\circ}, \beta = 94.6882^{\circ}, \gamma = 90^{\circ}$
Volume	$1389.36(10) \text{ Å}^3$
Z	4
Density (calculated)	1.182 g/cm^3
F(000)	536
Radiation type	Mo $K\alpha$
Theta range for data collection	2.4°–28.3°
Index ranges	$-11 \le h \le 11,$
	$-26 \le k \le 25,$
	$-10 \le l \le 10$
Tmin,Tmax	0.951 ;0.995
Reflections collected	41590
Independent reflections	4960[R(int) = 0.035]
Absorption correction	Multi-scan BRUKER SADABS2016/2
Refinement method	Full-matrix least-squares on F^2
Data/restraints/parameters	4960/0/139
Goodness-of-fit on F2	1.068
Δ/σ max	0.002
Refinement	$R[F^2 > 2\sigma(F^2)], wR(F^2), s 0.042, 0.107, 1.03$
Final R indices $I > 2\sigma(I)$	R1 = 0.035, wR2 = 0.0912
Final R indices all data	R1 = 0.1855, $wR2 = 0.1206$
Weighting scheme	$w = 1/[\sigma_2(F_o^2) + (0.044P)^2 + 0.5544P]$
Langest difference and half	where $P = (F_0^2 + 2F_c^2)/3$
Largest dill. peak and noie	$0.20 \text{ and } -0.23 \text{ eA}^{-3}$
K.M.S. deviation from mean	0.020 eA

Hydrogen-bond g	eometry in the crysta	l structure of com	pound 3.
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Hydrogen-bond geometry (Å, °)						
D - H - A	D - H	$H \cdots A$	$D \cdots A$	D - H - A		
N1—H1 <i>N</i> ···O2	0.855(17)	2.087(17)	2.7484(14)	133.8(14)		
N1—H1 N ···O2 ⁱ	0.855(17)	2.486(17)	3.1550(14)	135.8(14)		
C1— $H1C$ ···O2 ⁱⁱ	0.98	2.65	3.5084(17)	146		
Summatry adds: (i) $-y+1$ $-y+1$ $-z+1$: (ii) $-y+2$ $-y+1$ $-z+1$						

Symmetry codes: (i) -x+1, -y+1, -z+1; (ii) -x+2, -y+1, -z+1



Figure. S1. ¹H NMR (300 MHz) spectrum of compound 3.

Table S2



wavenumber (cm⁻¹)

Figure. S3. Experimental infrared spectrum of compound 3.



Figure. S4. Energy frameworks corresponding to the different energy components and the total interaction energy. (a) Energy framework diagram for separate electrostatic energy, (b) dispersion energy and (c) the total interaction energy components of the title molecule, results using CE-HF/3-21G (left side) and CE-B3LYP/6–31G(d,p) (right site). The energy factor scale is 100 and the cut-off is5.00kJ/mol.

				Table S2			
	Area percent report of GC.						
Peak	Retention time,	Area,	Height,	Area, %			
	min	$pA^{\cdot}s$	pA				
1	4.24139	3.77961	1.23719	0.037229			
2	4.70875	7.10704	1.86031	0.070005			
3	5.63347	1.67466	0.36761	0.016496			
4	6.02592	1285.30859	383.73224	12.66039			
5	6.67259	10.70552	2.35478	0.10545			
6	6.94303	536.85901	134.27452	5.288104			
7	8.37119	2.98387	0.40576	0.029391			
8	8.80731	6.41165	0.36927	0.063155			
9	9.58086	4.24333	0.67956	0.041797			
10	10.84589	8266.28613	466.50247	81.42358			
11	18.71644	14.98868	0.49225	0.14764			
12	20.27569	11.85334	0.40598	0.116756			