

OXAZIRIDINE (C-CH₃NO), C-CH₂NO RADICALS AND CL, NH₂ AND METHYL DERIVATIVES OF OXAZIRIDINE; STRUCTURES AND QUANTUM CHEMICAL PARAMETERS (Supplementary material)

Mohammad Taghi Taghizadeh*, Morteza Vatanparast, Saeed Nasirianfar

University of Tabriz, Faculty of Chemistry, 29, Bahman Blvd., Tabriz, 51666-14766, Iran

*e-mail: mttaghizadeh1947@gmail.com, mttaghizadeh@tabrizu.ac.ir, phone: (+98) 41 33 393 137; fax: (+98) 41 33 340 191

Supplementary material contains Tables S1 to S9 and Figures S1 to S4.

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Table S1

Structure of oxaziridine, radicals and their corresponding cations calculated using B3LYP/6-311++G (d, p) method [16].

<i>Oxaziridine</i>		<i>Radical 1</i>		<i>Radical 2</i>		<i>Radical 3</i>	
<i>Bond lengths (Å)</i>	<i>B3lyp</i>	<i>Ref [16]</i>	<i>B3lyp</i>	<i>B3lyp</i>	<i>B3lyp</i>	<i>B3lyp</i>	<i>B3lyp</i>
C1-O2	1.398	1.403	C1-O2	1.426	C1-O2	1.347	C1-O2
C1-H3	1.090	1.089	C1-H3	1.087	C1-H3	1.091	C1-H3
C1-H4	1.088	1.091	C1-H4	1.087	C1-N4	1.414	C1-N4
C1-N5	1.436	1.439	C1-N5	1.429	N4-H5	1.025	N4-H5
N5-H6	1.025	1.026	O2-N5	1.383	O2-N4	1.544	O2-N4
O2-N5	1.496	1.500					1.541
<i>Bond angles(°)</i>		<i>Bond angles(°)</i>		<i>Bond angles(°)</i>		<i>Bond angles(°)</i>	
H3-C1-O2	115.8		O2-C1-H3	115.6	H3-C1-O2	122.9	H3-C1-O2
O2-C1-H4	116.3	115.8	O2-C1-H4	115.7	O2-C1-N4	68.0	O2-C1-N4
N5-C1-H3	119.7	116.2	H3-C1-H4	117.3	C1-N4-H5	107.7	C1-N4-H5
C1-N5-H6	107.7	107.6	N5-C1-H3	118.3	O2-N4-H5	102.6	O2-N4-H5
O2-N5-H6	103.1		N5-C1-H4	118.3	N4-O2-C1	58.1	N4-O2-C1
H3-C1-H4	115.8		O2-C1-N5	58.0	C1-N4-O2	54.0	C1-N4-O2
N5-O2-C1	59.4		N5-O2-C1	61.1			
O2-C1-N5	63.7		C1-N5-O2	60.9			
C1-N5-O2	56.9						
<i>Torsion angles(°)</i>		<i>Torsion angles(°)</i>		<i>Torsion angles(°)</i>		<i>Torsion angles(°)</i>	
H4-C1-N5-H6	-157.5		N5-O2-C1-H3	108.7	H3-C1-N4-H5	-151.5	H3-C1-N4-H5
H3-C1-N5-H6	-11.67		N5-O2-C1-H4	-108.7			-23.2
<i>Oxaziridine Cation</i>		<i>Rad.1 Cation</i>		<i>Rad.2 Cation</i>		<i>Rad.3 Cation</i>	
<i>Bond lengths (Å)</i>	<i>B3lyp</i>	<i>Ref [16]</i>	<i>B3lyp</i>	<i>B3lyp</i>	<i>B3lyp</i>	<i>B3lyp</i>	<i>B3lyp</i>
C1-O2	1.502	1.507	C1-O2	1.529	C1-O2	1.239	C1-O2
C1-H3	1.085	1.086	C1-H3	1.086	C1-H3	1.092	C1-H3
C1-H4	1.084	1.088	C1-H4	1.086	C1-N4	1.338	C1-N4
C1-N5	1.416	1.439	C1-N5	1.442	N4-H5	1.037	N4-H5
N5-H6	1.032	1.032	O2-N5	1.223	O2-N4	1.677	O2-N4
O2-N5	1.311	1.317					1.677
<i>Bond angles(°)</i>		<i>Bond angles(°)</i>		<i>Bond angles(°)</i>		<i>Bond angles(°)</i>	
H3-C1-O2	113.3		O2-C1-H3	112.8	H3-C1-O2	136.0	H3-C1-O2
O2-C1-H4	114.0	113.3	O2-C1-H4	112.8	O2-C1-N4	81.1	O2-C1-N4
N5-C1-H3	119.0	113.9	H3-C1-H4	124.7	C1-N4-H5	115.8	C1-N4-H5
C1-N5-H6	140.3	140.3	N5-C1-H3	116.6	O2-N4-H5	106.4	O2-N4-H5
O2-N5-H6	122.1		N5-C1-H4	116.6	N4-O2-C1	52.0	N4-O2-C1
H3-C1-H4	121.9		O2-C1-N5	48.5	C1-N4-O2	46.9	C1-N4-O2
N5-O2-C1	56.0		N5-O2-C1	62.0			
O2-C1-N5	53.3		C1-N5-O2	69.5			
C1-N5-O2	66.7						
<i>Torsion angles(°)</i>		<i>Torsion angles(°)</i>		<i>Torsion angles(°)</i>		<i>Torsion angles(°)</i>	
H4-C1-N5-H6	-145.5		N5-O2-C1-H3	106.1	H3-C1-N4-H5	-96.2	H3-C1-N4-H5
H3-C1-N5-H6	14.1		N5-O2-C1-H4	-106.0			-96.2

Table S2

Atomic charges on C, N and O atoms from NBO calculation calculated using B3LYP level of theory and 6-311++G (d, p) basis set.

Structure	C	O	N
1	0.104	-0.419	-0.334
2	0.031	-0.344	-0.030
3	0.247	-0.382	-0.355
4	0.264	-0.384	-0.352
5	0.229	-0.394	-0.320
6	0.219	-0.395	-0.318
7	0.098	-0.358	-0.152
8	0.212	-0.347	-0.164
9	0.215	-0.360	-0.189
10	0.289	-0.391	-0.312
11	0.273	-0.362	-0.190
12	0.256	-0.429	-0.334
13	0.256	-0.429	-0.345
14	0.115	-0.424	-0.221
15	0.268	-0.436	-0.229
16	0.273	-0.437	-0.223
17	0.382	-0.439	-0.341
18	0.401	-0.453	-0.229
19	0.400	-0.446	-0.338
20	0.405	-0.428	-0.352
21	0.120	-0.507	-0.078
22	0.681	-0.450	-0.362
23	0.423	-0.660	-0.081
24	0.409	-0.567	-0.090
25	0.695	-0.687	-0.082

Table S3

HOMO and LUMO energy calculated using B3LYP/6-311++G(d,p).

Structure	HOMO	LUMO	$ \varepsilon_{HOMO} - \varepsilon_{LUMO} /(a.u)$	$ \varepsilon_{HOMO} - \varepsilon_{LUMO} /(eV)$
1	-0.27725	-0.01485	0.26240	7.14
2	α	-0.26441	0.25801	7.02
	β	-0.34121	0.24374	6.63
3	α	-0.23618	0.20002	5.44
	β	-0.29604	0.20682	5.63
4	α	-0.23490	0.19908	5.42
	β	-0.29651	0.21153	5.76
5	-0.30005	-0.03029	0.26976	7.34
6	-0.30595	-0.02904	0.27691	7.54
7	-0.29704	-0.08020	0.21684	5.90
8	-0.31679	-0.09387	0.22292	6.07
9	-0.30450	-0.08717	0.21733	5.91
10	-0.31627	-0.05155	0.26472	7.20
11	-0.31728	-0.09750	0.21978	5.98
12	-0.27033	-0.01281	0.25752	7.01
13	-0.27300	-0.01402	0.25898	7.05
14	-0.26317	-0.00845	0.25472	6.93
15	-0.25903	-0.00860	0.25043	6.81
16	-0.25189	-0.01155	0.24034	6.54
17	-0.26570	-0.01498	0.25072	6.82
18	-0.24766	-0.01233	0.23533	6.40
19	-0.25804	-0.01804	0.24000	6.53
20	-0.25814	-0.02235	0.23579	6.42
21	-0.27093	-0.02900	0.24193	6.58
22	-0.26037	-0.02011	0.24026	6.54
23	-0.23782	-0.08651	0.15131	4.12
24	-0.25659	-0.05476	0.20183	5.49
25	-0.23626	-0.08794	0.14832	4.04

Table S4

Selected natural bond orbital occupancies of oxaziridine and related radicals (a.u.) calculated using B3LYP/6-311++G (d, p) basis set. Only ring bonds (C-O, N-O and C-N) have been presented.

<i>Oxaziridine (structure 1)</i>		<i>Radical 1 (structure 2)</i>			
		<i>Spin</i>	α	β	
BD (1) C - O	1.98513	BD (1) C - O	0.99319	0.99300	
BD (1) O - N	1.97718	BD (1) O - N	0.99144	0.99134	
BD (1) C - N	1.98439	BD (2) O - N		0.97258	
BD*(1) C - O	0.01374	BD (1) C - N	0.99452	0.99424	
BD*(1) O - N	0.02295	BD*(1) C - O	0.00894	0.01014	
BD*(1) C - N	0.00802	BD*(1) O - N	0.00683	0.00624	
		BD*(2) O - N		0.00730	
		BD*(1) C - N	0.00345	0.00361	
<i>Radical 2 (structure 3)</i>		<i>Radical 3 (structure 4)</i>			
<i>Spin</i>	α	β	<i>Spin</i>	α	
BD (1) C - O	0.99240	0.99507	BD (1) C - O	0.99233	0.98646
BD (2) C - O		0.98108	BD (2) C - O		0.98261
BD (1) O - N	0.98739	0.98321	BD (1) O - N	0.98745	0.98255
BD (1) C - N	0.99088	0.98911	BD (1) C - N	0.99123	0.99145
BD*(1) C - O	0.00579	0.00562	BD*(1) C - O	0.00559	0.00861
BD*(2) C - O		0.04363	BD*(2) C - O		0.04443
BD*(1) O - N	0.01445	0.02413	BD*(1) O - N	0.01429	0.02355
BD*(1) C - N	0.00364	0.00424	BD*(1) C - N	0.00392	0.00354

Table S5

Selected natural bond orbital occupancies (a.u.) of compounds (structures 5-25) calculated using B3LYP/6-311++G (d, p) basis set. Only ring bonds (C-O, N-O and C-N) have been presented.

	Structure 5	Structure 6	Structure 7	Structure 8	Structure 9
BD (1) C - O	1.98624	1.98613	1.98420	1.98506	1.98530
BD (1) O - N	1.96892	1.96922	1.98121	1.97272	1.97513
BD (1) C - N	1.98344	1.98368	1.98628	1.98508	1.98529
BD*(1) C - O	0.04670	0.05145	0.01402	0.05537	0.05370
BD*(1) O - N	0.02277	0.02240	0.04655	0.05493	0.06143
BD*(1) C - N	0.03697	0.03714	0.01387	0.04906	0.06139
	Structure 10	Structure 11	Structure 12	Structure 13	Structure 14
BD (1) C - O	1.98781	1.98622	1.98185	1.98230	1.98166
BD (1) O - N	1.95967	1.96585	1.97570	1.97616	1.97175
BD (1) C - N	1.98324	1.98409	1.98038	1.98005	1.98017
BD*(1) C - O	0.08643	0.09434	0.03075	0.03229	0.01496
BD*(1) O - N	0.02071	0.06635	0.02365	0.02372	0.04600
BD*(1) C - N	0.06978	0.09901	0.02258	0.02260	0.01378
	Structure 15	Structure 16	Structure 17	Structure 18	Structure 19
BD (1) C - O	1.97894	1.97817	1.97905	1.97580	1.98453
BD (1) O - N	1.97063	1.97024	1.97441	1.97053	1.97379
BD (1) C - N	1.97629	1.97812	1.97625	1.97386	1.98273
BD*(1) C - O	0.03383	0.03198	0.04968	0.05103	0.07579
BD*(1) O - N	0.04681	0.04703	0.02372	0.04661	0.02329
BD*(1) C - N	0.02760	0.03011	0.03723	0.04394	0.04123
	Structure 20	Structure 21	Structure 22		
BD (1) C - O	1.98405	1.98364	1.98277		
BD (1) O - N	1.97361	1.96576	1.96992		
BD (1) C - N	1.98332	1.97633	1.98002		
BD*(1) C - O	0.04616	0.01478	0.10272		
BD*(1) O - N	0.02411	0.21708	0.02338		
BD*(1) C - N	0.06810	0.01355	0.08428		
	Structure 23		Structure 24		Structure 25
BD (1) C1 - O2	1.99510	BD (1) C1 - O2	1.98593	BD (1) C1 - O2	1.95925
BD (1) N4 - N5	1.98808	BD (1) O2 - N4	1.95523	BD (1) N3 - N10	1.98582
BD (2) N4 - N5	1.98367	BD (1) N4 - N5	1.99294	BD (2) N3 - N10	1.98158
BD (1) C1 - N4	1.96699	BD (1) C1 - N4	1.96393	BD (1) C1 - N3	1.96159
BD*(1) C1 - O2	0.05408	BD*(1) C1 - O2	0.06452	BD*(1) C1 - O2	0.06288
BD*(1) N4 - N5	0.07885	BD*(1) O2 - N4	0.31694	BD*(1) N3 - N10	0.10405
BD*(2) N4 - N5	0.31404	BD*(1) N4 - N5	0.00963	BD*(2) N3 - N10	0.27438
BD*(1) C1 - N4	0.13001	BD*(1) C1 - N4	0.04954	BD*(1) C1 - N3	0.15597

Table S6

Selected second order perturbation theory analysis of Fock Matrix in NBO Basis for structures calculated by B3LYP/6-311++G (d, p) method.
E (2) (kcal/mol) were reported.

Continuation of Table S6

<i>Donor NBO (<i>i</i>)</i>	<i>Acceptor NBO (<i>j</i>)</i>	<i>E</i> (2)	<i>Donor NBO (<i>i</i>)</i>	<i>Acceptor NBO (<i>j</i>)</i>	<i>E</i> (2)	<i>Donor NBO (<i>i</i>)</i>	<i>Acceptor NBO (<i>j</i>)</i>	<i>E</i> (2)
<i>I6</i>						<i>I8</i>		
BD (1) Cl - O2	BD*(1) O2 - N4	4.93	BD (1) C1 - O2	BD*(1) O2 - N4	4.03	BD (1) C1 - O2	BD*(1) O2 - N3	4.81
BD (1) O2 - N4	BD*(1) C1 - O2	5.60	BD (1) O2 - N4	BD*(1) C1 - O2	5.36	BD (1) O2 - N3	BD*(1) C1 - O2	5.73
BD (1) C5 - H6	BD*(1) O2 - N4	5.82	BD (1) C5 - H6	BD*(1) C1 - O2	5.09	BD (1) C4 - H6	BD*(1) O2 - N3	5.80
						BD (1) C8 - H9	BD*(1) C1 - O2	5.09
						BD (1) C12 - H13	BD*(1) C1 - N3	5.41
						BD (1) C12 - H14	BD*(1) C1 - O2	5.37
<i>I9</i>			<i>I0</i>			<i>I1</i>		
BD (1) Cl - O2	BD*(1) O2 - N5	4.35	BD (1) O2 - N5	BD*(1) C1 - O2	5.13	BD (1) C1 - O2	BD*(1) O2 - N5	3.80
BD (1) O2 - N5	BD*(1) C1 - O2	5.53	BD (1) O2 - N4	BD*(1) C1 - O2	5.13	BD (1) O2 - N5	BD*(1) C1 - O2	7.06
LP (1) N6	BD*(1) C1 - O2	15.00	LP (2) O2	BD*(1) C1 - H3	7.91	LP (1) N6	BD*(1) O2 - N5	36.59
			LP (2) O2	BD*(1) C1 - N6	7.54			
			LP (1) N4	BD*(1) C1 - N6	4.75			
			LP (1) N6	BD*(1) C1 - O2	4.06			
			LP (1) N6	BD*(1) C1 - N4	15.14			
<i>22</i>			<i>23</i>			<i>24</i>		
BD (1) Cl - O2	BD*(1) O2 - N4	3.69	BD (1) C1 - O2	LP (3) O2	7.77	BD (1) C1 - O2	BD*(1) O2 - N4	2.37
BD (1) O2 - N4	BD*(1) C1 - O2	5.11	BD (1) C1 - N4	LP (3) O2	8.53	BD (1) O2 - N4	BD*(1) C1 - O2	7.16
LP (1) N5	BD*(1) C1 - O2	17.23	BD (1) C1 - N4	BD*(1) N5 - H6	5.49	LP (1) N4	BD*(1) N5 - H7	7.65
LP (1) N8	BD*(1) C1 - N4	14.67	LP (3) O2	BD*(1) C1 - O2	15.76	LP (1) N5	BD*(1) O2 - N5	59.30
BD*(1) C1 - O2	BD*(1) C1 - N4	34.83	LP (3) O2	BD*(1) C1 - N4	19.75	LP (1) N8	BD*(1) C1 - O2	13.18
			LP (3) O2	BD*(2) N4 - N5	19.56	BD (1) C1 - N4	BD*(1) O2 - N4	5.28
			LP (1) N4	BD*(1) N5 - H7	8.18			
			LP (1) N8	BD*(1) C1 - N4	10.13			
			LP (1) N8	BD*(1) N5 - H7	6.61			
			BD*(2) N4 - N5	BD*(1) N4 - N5	23.69			
<i>25</i>								
LP (3) O2			BD*(1) C1 - O2	5.90				
LP (3) O2			BD*(1) C1 - N3	15.52				
LP (3) O2			BD*(1) N3 - N10	10.34				
LP (1) N4			BD*(1) C1 - O2	12.98				
LP (1) N7			BD*(1) C1 - N3	10.37				
LP (1) N7			BD*(1) N10 - H12	7.63				
BD*(2) N3 - N10			BD*(1) N3 - N10	41.17				

Table S7

**VIE and AIE (eV) for oxaziridine and three radicals calculated using B3LYP/6-311++G (d, p) method
(1Hartree=27.2114 eV) [16].**

IE B3LYP	This work			Reference [16]		
	VIE-AIE	IE		VIE-AIE	QCISD(T)	G2(MP2)
		B3LYP	QCISD(T)			
Oxaziridine	AIE	9.71	0.92	9.67	9.69	9.82
	VIE	10.63		10.61	10.65	10.76
Radical 1	AIE	9.86	0.64			
	VIE	10.50				
Radical 2	AIE	8.10	1.45			
	VIE	9.55				
Radical 3	AIE	8.03	1.50			
	VIE	9.53				

Table S8

Vertical Ionization Energies (eV) calculated using B3LYP/6-311++G (d, p).

Structure	VIE	Structure	VIE
1	10.63	14	9.93
2	10.50	15	9.66
3	9.55	16	9.49
4	9.53	17	9.95
5	10.87	18	9.23
6	11.09	19	9.72
7	10.74	20	9.69
8	11.06	21	9.94
9	10.73	22	9.67
10	11.10	23	8.73
11	10.93	24	9.29
12	10.22	25	8.58
13	10.31		

Table S9

**The value of the HOMO-LUMO energy (a.u.), Quantum chemical parameters:
Ionization potential (I), Electron Affinity (A), Electronegativity (χ), Chemical Potential (μ), Global Hardness (η),
Global Softness (S , σ) and Electrophilicity (ω); for oxaziridine and three radicals calculated
using B3LYP method and 6-311++G (d, p) basis set.**

Parameter	Oxaziridine	Radical 1	Radical 2	Radical 3
I=VIE (a.u.)	0.39 66	0.38587	0.35096	0.35016
A=VEA (a.u.)	0.02241	0.00114	0.00209	0.00399
χ (a.u.)	0.20653	0.19351	0.17652	0.17708
μ (a.u.)	-0.20653	-0.19351	-0.17652	-0.17708
η (a.u.)	0.18413	0.19237	0.17446	0.17309
S or σ (a.u.) ⁻¹	5.43103	5.19835	5.73277	5.77749
ω (a..)	0.11583	0.09732	0.08932	0.09058

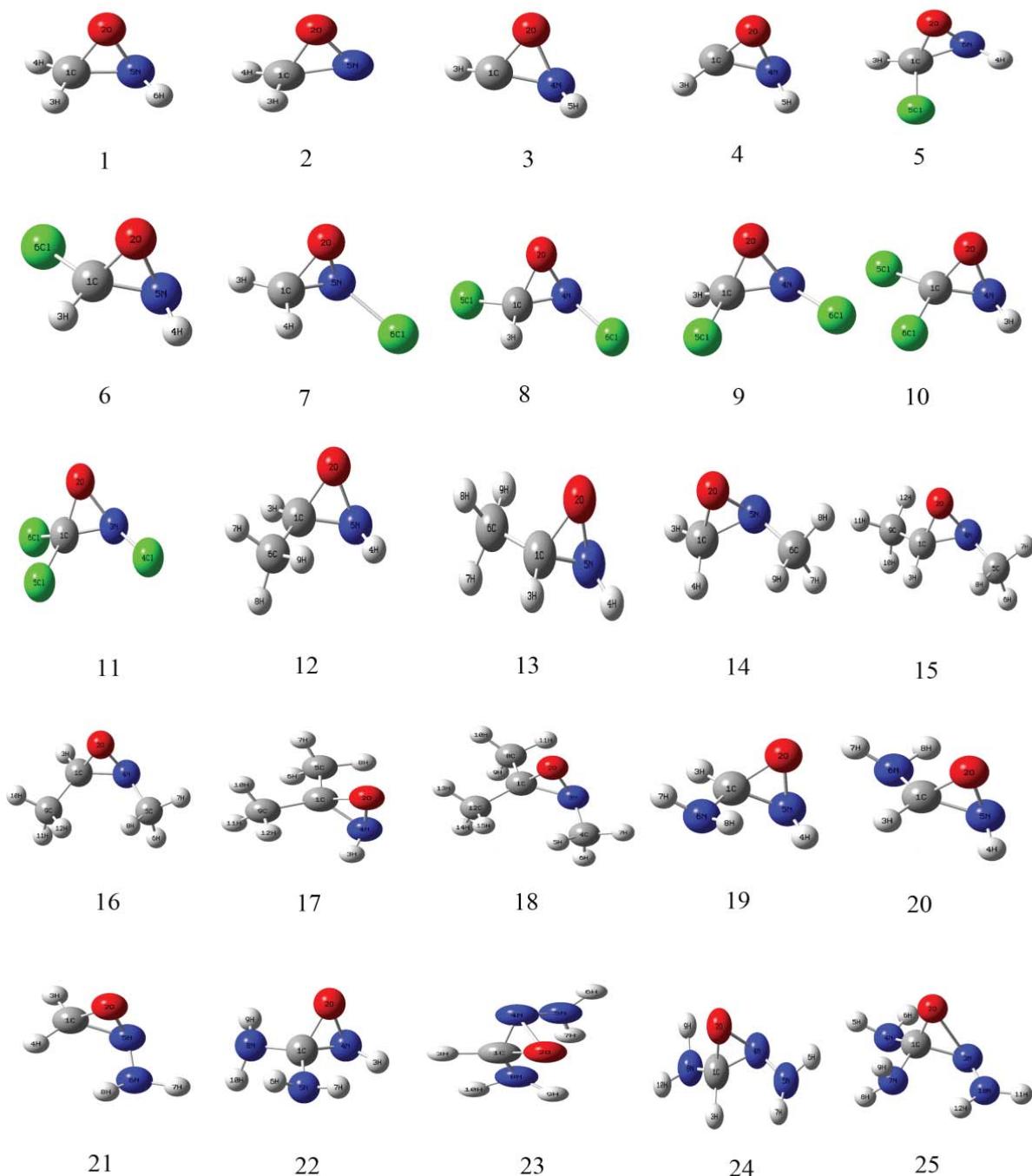


Figure S1. Optimum geometry of (1) oxaziridine [CH_3NO (${}^1\text{A}$)] (2) radical 1 [CH_2NO (${}^2\text{A}$)] (3) radical 2 [CH_2NO (${}^2\text{A}$)] (4) radical 3 [CH_2NO (${}^2\text{A}$)] (5) CH_2NOCl (${}^1\text{A}$) (6) CH_2NOCl (${}^1\text{A}$) (7) CH_2NOCl (${}^1\text{A}$) (8) CHNOCl , (${}^1\text{A}$) (9) CHNOCl , (${}^1\text{A}$) (10) CHNOCl_2 , (${}^1\text{A}$) (11) CNOCl_3 , (${}^1\text{A}$) (12) $\text{C}_2\text{H}_5\text{NO}$ (${}^1\text{A}$) (13) $\text{C}_2\text{H}_5\text{NO}$ (${}^1\text{A}$) (14) $\text{C}_2\text{H}_5\text{NO}$ (${}^1\text{A}$) (15) $\text{C}_3\text{H}_7\text{NO}$ (${}^1\text{A}$) (16) $\text{C}_3\text{H}_7\text{NO}$ (${}^1\text{A}$) (17) $\text{C}_3\text{H}_7\text{NO}$ (${}^1\text{A}$) (18) $\text{C}_4\text{H}_9\text{NO}$ (${}^1\text{A}$) (19) $\text{CH}_4\text{N}_2\text{O}$ (${}^1\text{A}$) (20) $\text{CH}_4\text{N}_2\text{O}$ (${}^1\text{A}$) (21) $\text{CH}_4\text{N}_2\text{O}$ (${}^1\text{A}$) (22) $\text{CH}_5\text{N}_3\text{O}$ (${}^1\text{A}$) (23) $\text{CH}_5\text{N}_3\text{O}$ (${}^1\text{A}$) (24) $\text{CH}_5\text{N}_3\text{O}$ (${}^1\text{A}$) (25) $\text{CH}_6\text{N}_4\text{O}$ (${}^1\text{A}$).

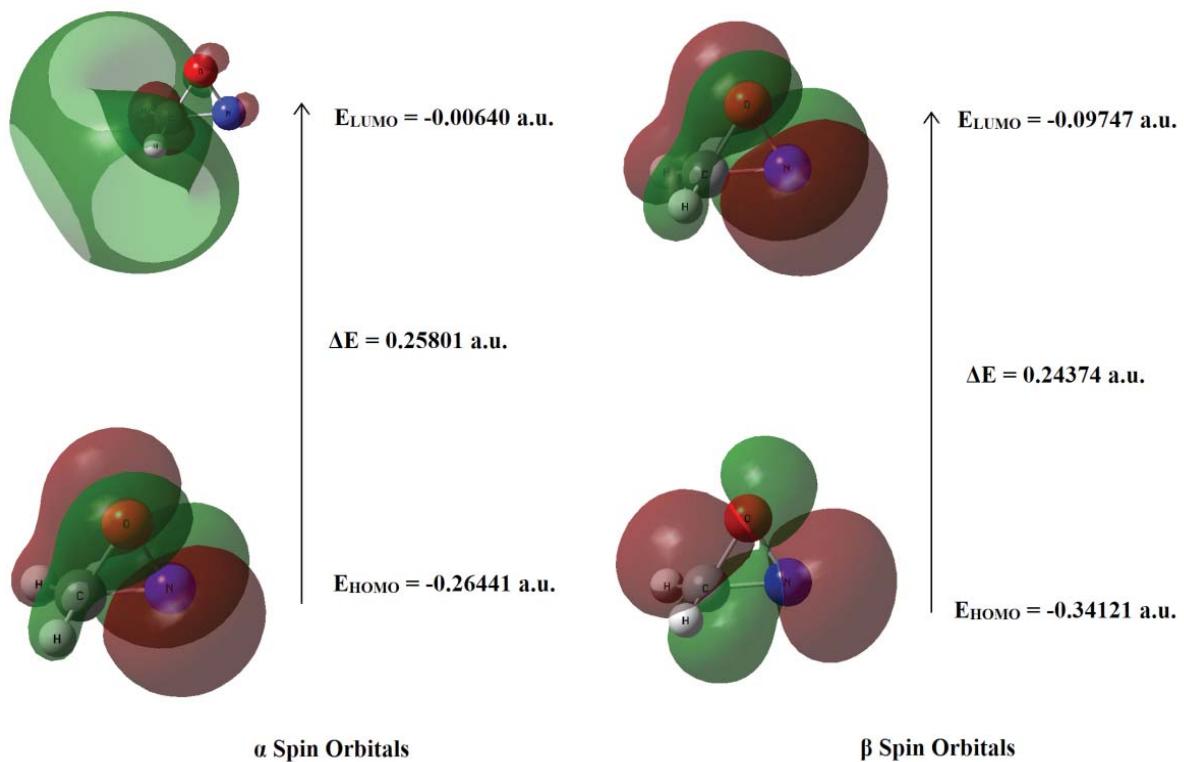


Figure S2. Isodensity plots of the frontier molecular orbitals of radical 1.

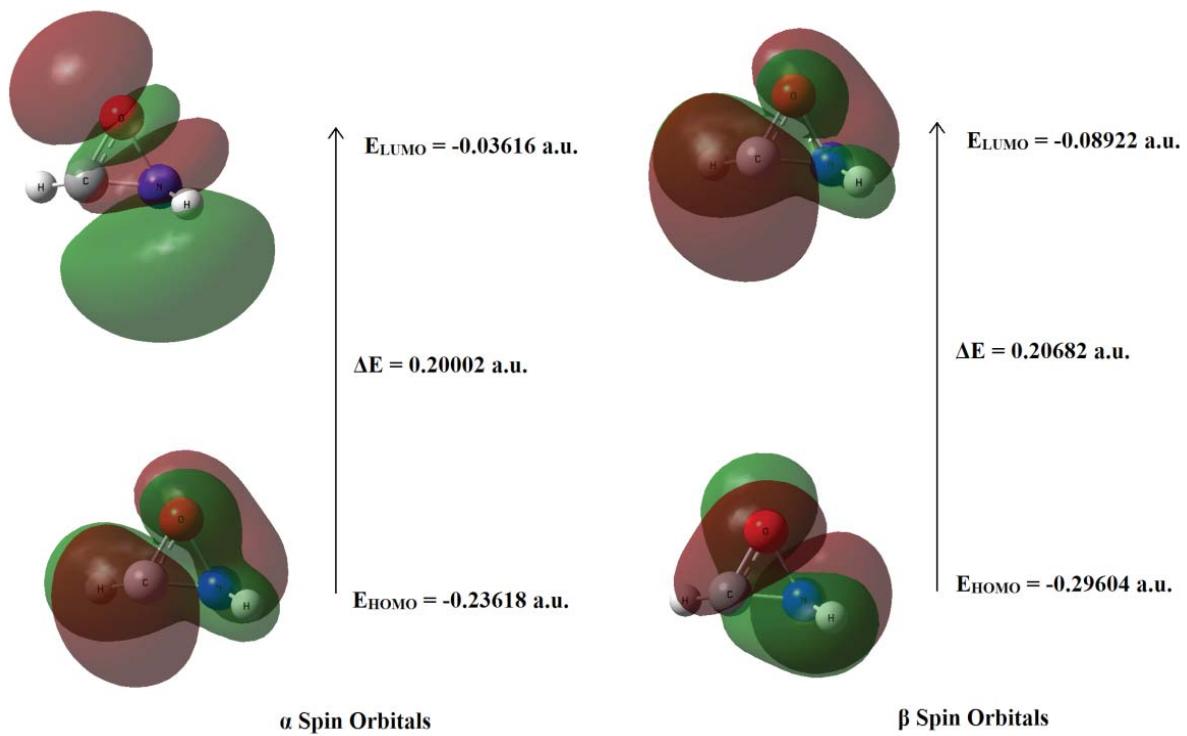


Figure S3. Isodensity plots of the frontier molecular orbitals of radical 2.

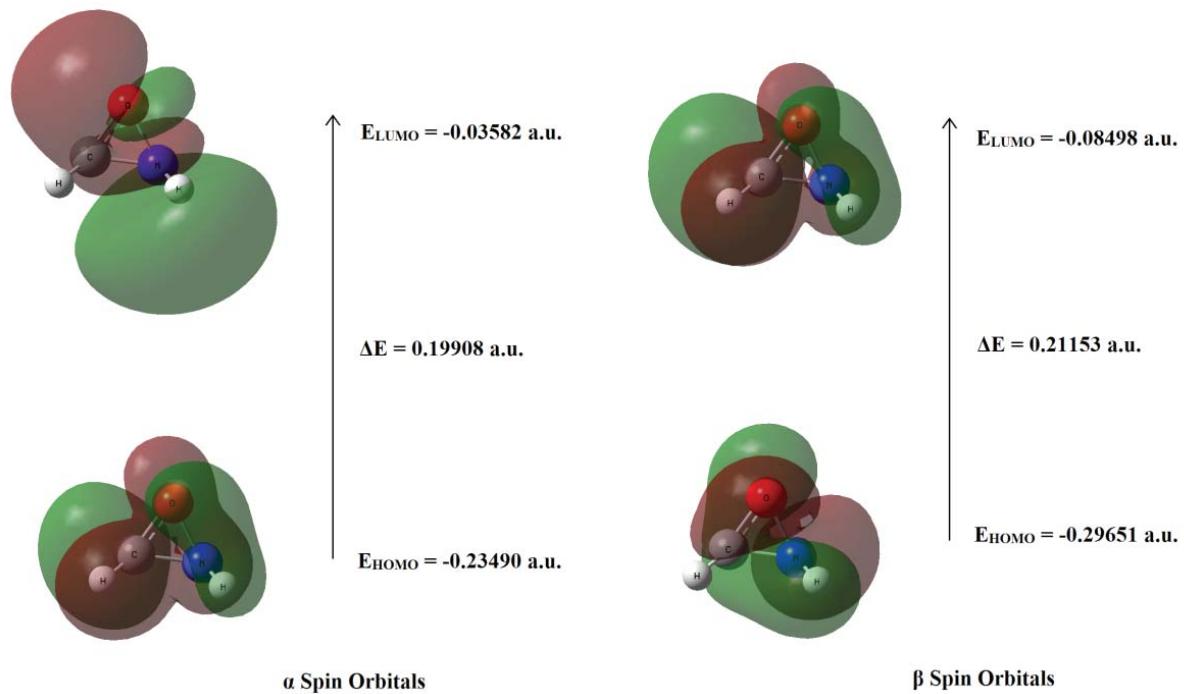


Figure S4. Isodensity plots of the frontier molecular orbitals of radical 3.